# S.P. Mandali's

# **RAMNARAIN RUIA AUTONOMOUS COLLEGE**



Syllabus for: S.Y. B.Sc.

Program: B.Sc.

Course Code: BIOTECHNOLOGY (RUSBTK)



Course code	Unit	Торіс	Credits	Lectures/
				week
Paper I: Biophysics	Unit I	Optics and Electromagnetic Radiations	2	1
RUSBTK301	Unit II	Heat, Sound, Magnetism and	_	1
		Fluid Dynamics		
	Unit III	Electrophoretic Techniques		1
Paper II :		Classical methods of analysis	2	1
Applied	Unit I		_	
Chemistry- I	Unit II	Environmental chemistry		1
RUSBTK302	Unit III	Green chemistry and nanomaterials		1
Paper III :	Unit I	Eff <mark>ectors o</mark> f Immune Response	2	1
Immunology	Unit II	Antigen-Antibody Interactions		1
RUSBTK303		Advanced Immuno-Techniques	-	1
	Unit III			
Paper IV :		Cytoskeleton	2	1
Cell Biology	Unit I			
and		Cell Membrane		1
Cytogenetics	Unit II		100	
RUSBTK304	Unit III	Cytogenetics		1
Paper V :		Gene Expression Transcription	2	1
Molecular	Unit I			
Biology RUSBTK305	ore	Gene Expression-	xce	1
	Unit II	Translation		
	Unit III			1
		<b>Regulation of Gene Expression</b>		
Paper VI:		Microorganisms in Industrial	2	1
Bioprocess		Processes		
Technology				
RUSBTK306				
	Unit I			
	Unit II	Fermentor and Fermentation Process		1

		]		
		Industrial Productions		1
	Unit III			
Paper VII :		Introduction to Research Methodology	2	1
Research		and Research Problem		
Methodology	Unit I			
		Research Design and Data		1
RUSBTK307				
		Collection		
	Unit II	Concention		
		Interpretation		1
		Interpretation		I
		and Report writing		
	Unit III			
Practicals	Practicals	2		
	based on			
	Paper I and			
	Paper II			
Practicals	Practicals	2		
	based on			
5	Paper III		1	
	and Paper			
	IV			
-	1			ſ
Practicals	Practicals	Evpori2hco	VCO	
	based on	Experiênce e E	VC	
	Paper V and	-		
	Paper VI			
	raper vi			
ТОТАТ		20		
TOTAL		20		
CREDITS				

Course code	Unit	Торіс	Credits	Lectures/ week
Paper I: Biochemistry RUSBTK401	Unit I	Carbohydrate Metabolism ETS	2	1
	Unit II	Amino Acid Metabolism		1
	Unit III	Lipid Metabolism		1
Paper II : Applied Chemistry- II RUSBTK402	Unit I	Physical Chemistry	2	1
	Unit II	Sepa <mark>ration</mark> techniques in analytical chemistry		1
	Unit III	UV-Visible absorption spectroscopy		1
Paper III : Medical Microbiology		Infectious Diseases	2	1
RUSBTK403	Unit I	Infectious Diseases		
RI	Unit II	Medical Microbiology- Causative Organisms- I	G	1
	Unit III	Medical Microbiology - Causative Organisms- II		1
Paper IV : Environmental Biotechnology	'e •	Experience		cel
RUSBTK404	Unit I	Renewable sources of energy		
	Unit II	Xenobiotic compounds and waste water treatment		1
	Unit III	Bioremediation		1
Paper V : Biostatistics and Bioinformatics RUSBTK405	Unit I	Introduction to Computers and Biological Databases	2	1

		BLAST & Structural		1
		Bioinformatics		1
	Unit II	Diomiormatics		
	Unit II	<b>D!</b> ( ) ( )		1
		Biostatistics		1
	Unit III			
Paper VI:			2	1
Molecular				
Diagnostics				
		<b>Basics of Molecular</b>		
RUSBTK406	Unit I	Diagnostics		
		Nucleic Acid Amplification		1
		and hybridisation Methods		
	Unit II			
		Molecular Biology based		1
		Diagnostics		
	Unit III			
Paper VII :			2	1
-		Entropyonourship	4	1
Entrepreneurship		Entrepreneurship		
Development	<b>T</b> T •/ <b>T</b>	Development		
RUSBTK407	Unit I			
		Setting-up of an Business		1
		enterprise & management		
		aspect		
	Unit II		1000	
		Innovation and		1
		Entrepreneurship		7
	Unit III	development		
Practicals	Practica	ls based on Paper I and Paper	2	
<b>FYNO</b>	'O O	- ymorionco		
FVDIA		PYDELICITCE		
Practicals	Practical	s based on Paper III and Paper	2	
		IV	_	
		<b>A V</b>		
Practicals	Practical	ls based on Paper V and Paper	2	
			4	
		VI		
TOTAL ODDDIEG			20	
TOTAL CREDITS			20	

# S.Y.BSc BIOTECHNOLOGY (2019-2020) SEMESTER III Paper I-Biophysics

### Course objectives: -

The objective of this course is to have a firm foundation of the fundamentals and applications of current biophysical theories.

Learning outcomes: - By the end of the course the student will:

- □ Develop an understanding of the different aspects of classical Physics.
- □ Be able to relate principles of Physics to applications and techniques in the field of Biology such as Microscopy, Spectroscopy and Electrophoresis

RUSBTK301Introduction to Optics and Lasers: Optics : Properties of Light Reflection, Refraction, Dispersion, Interference. Laser : Electromagnetic Radiations: Introduction to Electromagnetic RadiationsIntroduction to Electromagnetic Radiations: Introduction to Electromagnetic Radiations: Spectroscopy : Types and Properties of Spectra; Basic Laws of Light Absorption. SpectrophotometerPrinciple, Instrumentation and Applications; UV-Vis Spectrophotometer. Microscopy: Types of Microscopy; Electron Optics; Electron Microscopy.15	Course Code	UNIT	TOPICS	Credits	Lectures
RUSBTK301       Optics : Properties of Light - Reflection, Refraction, Dispersion, Interference.         Lasers : Properties of Lasers, Stimulated Emissions, Laser Action; Applications of Laser.       Progetties of Laser.         Electromagnetic Radiations: Introduction to Electromagnetic Radiation.       Spectroscopy : Types and Properties of Spectra; Basic Laws of Light Absorption.       15         Spectroscopy : Types and Applications; UV-Vis Spectrophotometer, Single and Dual Beam Spectrophotometer, Single and Immuno-Electron Microscopy.					
	RUSBTK301	Optics Electromagn	Optics :Properties of LightReflection, Refraction,Dispersion, Interference.Lasers :Properties of Lasers,Stimulated Emissions, LaserAction; Applications ofLaser.Electromagnetic Radiations:Introduction to ElectromaRadiation.Spectroscopy :Introduction to ElectromaRadiation.Spectroscopy :Microscopy:Types and Properties ofSpectrophotometer,Single and DualSpectrophotometer.Microscopy:Types of Microscopy;Electron Optics; ElMicroscopy.Preparation of Specimen,TEM and Immuno-ElMicroscopy.	Ignetic 2 V-Vis Beam lectron SEM,	15 cel
UNIT II Heat: 15		UNIT II	Heat:		15

Heat, Sound,	Concept of Temperature; Modes		]
Magnetism and	of Heat Transfer; Measuring		
Fluid Dynamics	Temperature; Platinum Resistance		
i fuit Dynamics	Thermometer; Thermocouple and		
	Thermistors.		
	Sound:		
	Types of Sound Waves		
	Audible, Ultrasonic and		
	Infrasonic Waves; Doppler		
	Effect; Applications of		
	Ultrasonic Waves.		
	Magnetism:		
	Magnetic Field; Magnetism		
	of Earth; Para-magnetism,		
	Diamagnetism,		
	Ferromagnetism. Nuclear		
	Magnetism and		
	Biomagnetism.		
	Fluid Dynamics :		
	Viscosity:		
	Definition Flow of Liquids through		
	Capillaries; Stokes' Law; Terminal		
	Velocity. Determination of $\eta'$		
	by Falling Sphere Method; Viscosity		
	Estimation by Oswald's Viscometer.		
	Surface Tension:		
	Definition - Surface Tension and		
	Surface		
	Energy; Capillary Action; Angle of		
	Contact;		
	Wettability; Temperature		
	Dependence of Surface Tension.		
	Applications in Biology.		
	Electricity- AC & DC		
EVID DKO	Electrophoresis:		
CXDIDIE 🔵	Migration of Ions in an applied	D CX(	e
	electric field; Factors affecting		- 10 M
Electrophoresis	Electrophoretic Mobility; Moving		
	Boundary Electrophoresis; Paper		
	Electrophoresis; AGE; Native and		
	SDS PAGE (reducing and		1 =
	nonreducing, continuous and		15
	discontinuous); IEF and 2D PAGE,		
	Pulse field and Capillary		
	electrophoresis. Staining and		
	Detection Methods; Gel –		
	Documentation, Applications in		
	Biology		
	Бююбу		

# Paper II – Applied chemistry

#### Course objectives: -

The objective of this course is to have a firm foundation of the fundamentals and applications of Gravimetric analysis and Green Chemistry.

**Learning outcomes:** - By the end of the course the student will be able to:

- □ Scope and importance of analytical chemistry.
- $\Box$  Concept of accuracy and precision.
- □ Chemistry of water and various aspects of assessment of quality of water.

Course	UNIT	TOPICS	Credits	Lectures
Code		101105	ereans	Lectures
Cout		Gravimetric analysis:	2	15
	UNIT I	Introduction to gravimetric	2	13
RUSBTK302	Classical methods of analysis	analysis, types of gravimetric analysis, conditions for a reaction to be used in gravimetric analysis, solubility and solubility product, factors affecting solubility: temperature, common and diverse ion effect, pH, nature of the solvent, complexation.		
		Unit operations in gravimetric analysis, precipitation, homogenous and heterogeneous precipitation, relative super saturation, nucleation and crystal growth, their effect on particle size, Ostwald's ripening, impurities associated with precipitate formation, filtration,		
R	RUIA	precipitate formation, filtration, washing of the precipitate, drying and incineration, use of thermal methods.		
Exp	lore •	Titrimetric analysis Introduction to titrimetric analysis, conditions for a reaction to be used in titrimetric analysis, terms involved: titrant, titrand, indicator, equivalence point, endpoint, titration error, types of titrations.	• E	xcel
		Acid –base titrations Acid base indicators, theory of acid base indicators, conditions for choosing an indicator. Types of acid base titrations, titration curves. Construction of the titration curves and the choosing of the indicator for		

	A) strong acid –strong base B) strong acid –weak base C) weak acid – strong base D) weak acid – weak base Titration of dibasic acid with a strong base, condition for obtaining two separate equivalence points, qualitative description of the titration curve, determination of the dissociation constant Titration of phosphoric acid with a strong base	
UNIT II Environmental Chemistry	Chemistry of water Water as a natural resource: Physical and Chemical properties of water, significance of water as a universal solvent and its properties viz. pH, Dielectric constant, boiling point. Anomalous behaviour of water. Hydrological cycle. Chemical composition of ground water. Factors affecting solubility of gases in water. Solubility of CO <sub>2</sub> and O <sub>2</sub> in water Water quality : Parameters for determining water quality i) Physical parameters: - pH, pE, conductivity, TS, TSS, TDS ii) Chemical Parameters- acidity, alkalinity, hardness, salinity , chlorine demand , DO, COD, iii) Biological parameter BOD MPN	
UNIT III Green Chemistr & Nanomaterials	<ul> <li>Biological parameter – BOD, MPN</li> <li>Standards for Potable and industrial water.</li> <li>Green Chemistry and Synthesis: Introduction to Green Chemistry; Need and Relevance of Green Chemistry; Principles of Green Chemistry.</li> <li>Green Synthesis in Industry: Green Materials, Green Reagents, Green Solvents and Green Catalysts.</li> <li>Nanomaterials: Introduction to Nanomaterials.</li> <li>Forms of Nanomaterials: Nanoparticles, Nanofilms and Nanotubes Synthesis and Characterization of Nanomaterials.</li> </ul>	• E scel

	Applications Nanomaterials.	of	

# Paper III- Immunology

and various Imr Learning outco Understand Mechanisms Understand	munotechniqu omes: - By th the role of di s in Immunol	e end of the course the student will be fferent types of Cells, Effector Molecu ogy.	able to:		
Learning outco	omes: - By th the role of di s in Immunol	e end of the course the student will be fferent types of Cells, Effector Molecu ogy.		tor	
<ul> <li>Understand</li> <li>Mechanisms</li> <li>Understand</li> </ul>	the role of di s in Immunol	fferent types of Cells, Effector Molecu ogy.		tor	
<ul> <li>Understand</li> <li>Mechanisms</li> <li>Understand</li> </ul>	the role of di s in Immunol	fferent types of Cells, Effector Molecu ogy.		tor	
Mechanisms Understand	s in Immunol	ogy.			
□ Understand					
		s underlying various Immuno-techniqu	es.		
Course	UNIT	TOPICS	Credits	Lectures	
Code					
	UNIT I	Haematopoiesis; Complement System-	2	15	
	Effectors of	Classical, Alternate and Lectin;			
	Immune	Regulation and Biological Effects of			
	Response	Complement System; Deficiencies of			
		Complement System			
DI LODTI (O O O	UNIT II	Antigen antibody interaction		15	
	Antigen	techniques- Precipitation Reactions:			
	antibody	Immunoprecipitation,			
	interaction	Immunoelectrophoresis, CIEP, Rocket			
		Electrophoresis and 2-D			
		Immunoelectrophoresis			
		Agglutination Reactions:			
		Passive, Reverse Passive,			
		Agglutination Inhibition.			
	5667	Coomb's Test; Complement Fixation		and the second second	
P		Tests.			
		Synthesis of Monoclonal antibodies &			
	Applications.				
	Unit III	RIA, ELISA, ELISPOT,		15	
	Advanced	Chemiluminescence, Western Blot,			
	Immuno-	Immunofluroscence, Flow Cytometry.		YCOL	
PVA	Techniques	Alternatives to Antigen- Antibody Reactions		VCI	

#### **SEMESTER III**

#### Paper IV - CELL BIOLOGY AND CYTOGENETICS

#### Course objectives: -

The objective of this course is to have a firm foundation in the fundamentals of Cell Biology and Cytogenetics.

**Learning outcomes:** -By the end of the course the student will be able to:

- Develop an understanding of the Cytoskeleton and Cell Membrane.
- □ Discuss the structure of Chromosomes and types of Chromosomal Aberrations.
- □ Discuss the principles underlying Sex Determination, Linkage and Mapping.

Course	UNIT	TOPICS	Credits	Lectures
Code				
Code RUSBTK304	UNIT I Cytoskeleton	Cytoskeleton: Overview of the Major Functions of Cytoskeleton. Microtubules: Structure and Composition. MAPs: Functions- Role of Mitosis, Structural Support and Cytoskeleton Intracellular Mobility. Motor Proteins: Kinesins, Dynein; MTOCs. Dynamic Properties of Microtubules. Microtubules in Cilia	2	15
<b>Exp</b>	UNIT II Cell Membrane	and Flagella. Microfilaments: Structure, Composition, Assembly and Disassembly. Motor Protein: Myosin. Muscle Contractility: Sliding Filament Model. Actin Binding Proteins; Examples of Non Muscle Motility. Intermediate Filaments: Structure and Composition; Assembly and Disassembly; Types and Functions Cell Membrane: Uptake of Nutrients by Prokaryotic Cells; Overview of membrane functions, History on Plasma membrane structure, Chemical composition of membranes-Membrane carbohydrates & Membrane lipids. Integral, Peripheral & Lipid anchored membrane proteins, Importance & maintenance of Membrane fluidity, Asymmetry of membrane lipids, Lipid rafts, Diffusion of Membrane proteins after cell fusion, Restrictions on Protein and Lipid mobility, Red Blood cell – An example of Plasma membrane structure.	GE O Ex	15 Cel

UNIT III	Cytogenetics:	15
Cytogenetics	Structure of Chromosome-	
	Heterochromatin, Euchromatin,	
	Polytene Chromosomes.	
	Variation in Chromosomal Structure	
	and Number:	
	Deletion, Duplication, Inversion,	
	Translocation, Anueploidy, Euploidy	
	and Polyploidy and Syndromes-	
	Klienfelter, Turner, Cri-du-chat,	
	Trisomy -21, Trisomy 18 and Trisomy	
	13.	
	Sex Determination and Sex Linkage:	
	Mechanism of Sex Determination	
	(XX-XY, ZZ-ZW, XX-XO)	
	Dosage Compensation and Barr Body.	
	Genetic Linkage, Crossing Over and	
	Chromosomal Mapping:	
	Tetrad Analysis, Two-point Cross,	
	Three Point Cross, Pedigree Analysis	

# SEMESTER III

## Paper V – Molecular Biology

Course objectives: -

The objective of this course is to have an insight into mechanism of Gene Expression and Regulation.

Learning outcomes: -By the end of the course the student will be able to:

- Discuss the mechanisms associated with Gene Expression at the level of Transcription and Translation.
- Discuss the mechanisms associated with Regulation of Gene Expression in Prokaryotes and Eukaryotes.

Lukaryotes				
Course	UNIT	TOPICS	Credits	Lectures
Code				
	UNIT I	Gene Expression- an Overview.	2	15
1.1.1	Gene	Transcription Process in		
<b>FVN</b>	Expression –	Prokaryotes :		<b>ICO</b>
FVA	Transcription	RNA Synthesis; Promoters and		
		Initiation of Transcription at		
		Elongation and Termination of		
		Chain.		
RUSBTK305		Transcription in Eukaryotes		
		Transcription of Protein Coding		
		Genes by RNA Polymerase		
	UNIT II	Nature of Genetic Code.		15
	Gene	Wobble Hypothesis.		15
	Expression-	Translation:		
	Translation	Process of Protein Synthesis		
		(Initiation, Elongation,		
		Translocation, Termination)		
		. , , , , , , , , , , , , , , , , , , ,		

UN	NIT III	In prokaryotes:	15
Re	egulation of	In Bacteria:	
Ge	ene	Lac operon of E.coli, trp Operon of	
Ex	xpression	E.coli.	
		In Eukaryotes:	
		Operons in Eukaryotes; Control of	
		Transcriptional Initiation	
		Jumping genes in maize	

# SEMESTER III Paper VI- Bioprocess technology

### Course objectives: -

The objective of this course is to understand the basics skills applied in Fermentation Technology and build a foundation for more advanced studies in Bioprocess Technology. Learning outcomes: -By the end of the course the student will be able to:

- Develop an understanding of the various aspects of Bioprocess Technology.
- Develop skills associated with screening of Industrially Important Strains.
- □ Understand principles underlying design of Fermenter and Fermentation Process.

Course	UNIT	TOPICS	Credits	Lectures
Code				
	UNIT I	Types of Microorganisms	2	15
	Microorganisms	used in Industrial		
	in Industrial	Processes :		
	Processes	Bacteria, Fungi		
RUSBTK306	RUIA	Screening and Maintenance of Strains: Primary Screening and Secondary	GE	
		Screening; Cultivation; Preservation of		
Exp	ore •	Cultivation; Preservation of Industrially Important Microbial Strains. Strain improvement, Inoculum development – One example each.	<b>Exc</b>	el
	UNIT II	<b>Design of a fermentor:</b> Stirred Tank		15
	Fermentor and	Fermentor, Air lift, Pneumatic,		
	Fermentation Processes	Bubble column, Tower fermentor, - Basic Design; Parts of a Typical Industrial Fermentor. Process Parameters :		
		<i>pH</i> , Temperature, Aeration,		
		Agitation, Foam, Pressure, Inlet and exit gas analysis, Dissolved oxygen. Carbon dioxide electrodes, microbial biomass, Safety		

UNIT III Industrial Productions	valves.  Product Isolation and Purification. Study of Representative Fermentation Processes : Outline of Penicillin, Ethanol, Streptomycin & Vinegar Production by Fermentation along with a flow diagram	15
_	flow-diagram.	

# SEMESTER III Paper VII – Research Methodology

**Course objectives: -**The objective of this course is to develop Research Aptitude, Logical Thinking and Reasoning.

Learning outcomes: -By the end of the course the student will be able to:

Understand basic principles of Research Methodology and identify a Research Problem.

□ Understand a general definition of Research Design.

□ Identify the overall Process of Designing a Research Study from its inception to its Report.

Course	UNIT	TOPICS	Credits	Lectures
Code				
RUSBTK307	UNIT I Introduction to Research Methodology and Research Problem	Meaning of Research; Objectives of Research; Motivation in Research; Types of Research; Research Approaches; Significance of Research; Research Methods versus Methodology; Research Process; Criteria of Good Research; Problems Encountered by Researchers in India; What is a Research Problem? Selecting the Problem; Necessity of Defining the Problem; Technique Involved in Defining a Problem		<sup>15</sup> xcel
	UNIT II Research Design And Data Collection	Meaning of Research Design; Need for Research Design; Features of a Good Design; Important Concepts Relating to Research Design; Different Research Designs; Basic Principles of		15

UNIT III Interpretation and Report Writing	Experimental Designs; Developing a Research Plan- Collection of Primary Data; Observation Method; Interview Method; Collection of Data through Questionnaires; Collection of Data through Schedules; Other Methods of Data Collection, Collection of Secondary Data, Selection of Appropriate Method for Data Collection, Case Study Method Meaning of Interpretation, Why Interpretation?, Technique of Interpretation, Significance of Report Writing, Different steps in report writing, Layout of research report, Types of report, Mechanics of writing a research report, Precaution for writing a research report.	15



# RUIA COLLEGE Explore • Experience • Excel

Course Code	Title	Credits
Course Code Based on Paper 1 and 2 RUSBTKP301	<ul> <li>Determination of Purity of Plasmid DNA using UV Spectrophotometry.</li> <li>Study of the Structure and Function of an Electron Microscope (Visit / Video Demonstration - including Sample Preparation and Staining).</li> <li>Demonstration of Structure and Working of a Fluorescence Microscope (Stained Preparation).</li> <li>Electrophoresis of Proteins by PAGE.</li> <li>Chemical and Biological Synthesis of Silver Nanoparticles and its Characterisation by UV- VIS Spectrophotometer</li> <li>To study the kinetics of the reaction between K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> and KI for equal concentration.</li> <li>To determine the amount of strong acid in the given solution by conductometric titration.</li> <li>To determine the amount of strong acid in the given solution by pH-metric titration.</li> <li>Organic preparation and their purification: Use 0.5-1.0g of the organic compound. Purify the product by recrystallization. Report theoretical yield, percentage yield and melting point of the purified product. Preparation of:</li> <li>a. m-Dinitrobenzene from nitrobenzene</li> <li>b. Phthalic anhydride from phthalic acid by sublimation</li> <li>c. P-bromoacetanilide from acetanilide</li> <li>Quantitative determination of salts such as copper sulphate pentahydrate, nickel chloride hexahydrate, anhydrous cupric chloride using standard volumetric methods (any 1)</li> <li>Gravimetric estimation of Nickel (II) as Ni-DMG.</li> </ul>	Credits 2
Course Code	Title COLEGE	Credits
Based on Paper 3 and 4 RUSBTKP303	<ul> <li>Passive Agglutination- RA Factor Test.</li> <li>ELISA (Kit based).</li> <li>Dot ELISA</li> <li>Single radial immunodiffusion</li> <li>Ouchterlony double immunodiffusion</li> <li>Study of Normal and Abnormal Karyotype (Chromosomal Aberration) - Deletion, Duplication, Inversion, Translocation and Syndromes- Trisomy 21, Trisomy 13, Trisomy 18, Klienfelter, Turner and Cri-du-Chat using Software.</li> <li>Video demonstration of G- banding</li> <li>Induction of Polyploidy by PDB/ Colchicine/ UV Treatment Using Suitable Plant material</li> <li>Study of Polytene Chromosomes</li> <li>Mapping based on Tetrad Analysis and Three Point Cross.</li> <li>Pedigree Analysis- Autosomal and Sex- Linked.</li> </ul>	2 el

# Practical for semester III

	• Study of E.coli Diauxic Growth Curve- (Lactose and Glucose)	2
	• Expression of β- galactosidase and Measurement of Activity.	
	• Screening for an Antibiotic Producing Strain of Microorganism	
	• Screening for an Acid producing strain of microorganism	
	Screening for an Alcohol Producing Strain of Microorganism	
	• Lab Scale Production of Penicillin (Static and Shaker)	
Based on	Lab Scale Production of Ethanol	
Paper 5and 6	• Estimation of <i>Penicillin</i> from Recovery Broth by Chemical	
_	(Iodometric) Method.	
	• Estimation of <i>Penicillin</i> from Recovery Broth by Biological	
RUSBTKP305	(Bioassay) Method.	
	Estimation of Vinegar	
	Estimation of Alcohol from Recovered Broth by Dichromate	
	Method.	
	• Isolation, Quantitative Analysis and AGE of Genomic	
	DNA from Bacteria and Yeast.	



# RUIA COLLEGE Explore • Experience • Excel

# SEMESTER IV Paper I - Biochemistry

Course objectives: -The objective of this course is to gain an insight into the Metabolic Processes associated with Catabolism of Carbohydrates, Amino Acids, Lipids and Nucleotides. Learning outcomes: -By the end of the course the student will be able to Discuss the Metabolic Pathways of Carbohydrates, Amino Acids, Lipids and Nucleotides. Course UNIT TOPICS Credits Lectures Code 2 UNIT I **Carbohydrate Metabolism:** 15 Carbohydrate Glycolytic Pathway and its Metabolism. Regulation, Homolactic ETS Fermentation; Alcoholic Fermentation; Energetics of Fermentation; Citric Acid Cycle and RUSBTK401 its Regulation **Electron Transport System:** Electron Transport and Oxidative Phosphorylation. Inhibitors of ETS UNIT II Amino Acid Breakdown: 15 Amino Acid Deamination, Transamination, Urea Cycle, Breakdown of Glucogenic and Metabolism Ketogenic Amino Acids. Amino Acids as Biosynthetic Precursors. Unit III Lipid Metabolism: 15 Lipid Mobilization, Transport of Fatty Metabolism Acids. Beta, Alpha and Omega Oxidation of Saturated Fatty Acids; Oxidation of Unsaturated Fatty Acids, Oxidation of Odd Chain Fatty Acids. Energy Yield, Ketone Body Breakdown to Yield Energy. (Sequence of Reactions, Regulation, Energy Yield and Metabolic Disorders of the above Pathways)

# Paper II – Applied chemistry

#### Course objectives: -

The objective of this course is to have a firm foundation of the fundamentals and applications of current Physical chemistry.

**Learning outcomes:** -By the end of the course the student will:

- □ Types of electrodes and electrochemical cells.
- $\Box$  Nernst equation and its importance.
- □ Calculation of pH for strong and weak electrolytes.

Course	UNIT	TOPICS	Credits	Lectures
Code				
Code RUSBTK402	Unit I Physical Chemistry	Electrochemistry: Electromotive Force of Galvanic Cells Electrochemical cells, galvanic cells, reversible cells and reversible electrodes, conventions to represent Galvanic cells. Types of electrodes, standard electrode potential, electrochemical series. Cell potential and standard cell potential. Nernst equation and its importance. Calculation of thermodynamic parameters: $\Delta G$ , $\Delta H$ , $\Delta S$ and equilibrium constant from EMF data. Classification of galvanic cells: chemical cells and concentration	2	15
Expl	UNIT II Separation techniques in analytical chemistry	chemical cells and concentration cells Determination of pH using glass electrode and quinhydrone electrode. <b>pH and Buffers</b> pH concept, calculation of pH for strong and weak electrolytes Buffer, Henderson's equation for acidic and basic buffer Buffer Capacity. Numerical Problems based <b>Introduction to separation</b> <b>Techniques</b> Separation and its importance in analytical chemistry, estimation without separation. Classification of separation methods physical and chemical	G	E xcel

	complex formation.	
	Physical methods of separation,	
	precipitation, fractional	
	precipitation, volatilization,	
	distillation, fractional distillation,	
	vacuum distillation.	
	Solvent extraction	
	Nernst's distribution law, partition	
	coefficient, distribution ratio,	
	Percentage extraction, extraction	
	efficiency, percentage extraction	
	for single step and multistep	
	process with the same total volume	
	of the extracting solvent	
	Modes of extraction: Chelation,	
	ion-pair formation and solvation.	
	Batch and continuous extraction,	
	Counter current extraction	
	Chromatography	
	Introduction, Stationary and	
	mobile phase, common features of	
	all chromatographic techniques,	
	classification of chromatographic	
	methods on the basis of physical	
	state of the two phases.	
	Denon almometry and	
	Paper chromatography	
	Introduction and basic principles.	
	Stationary phase, transfer of the	
	sample, mobile phase.	
	Methods of developing the	
	chromatogram, methods of	
P. J	detection, physical, chemical and	
	enzymatic.	
	Applications.	
	Comparison of the paper and thin	
-	layer techniques.	100
LYB AKA	Eventing -	LVCA
	Thin layer chromatography	COXICE
	Introduction, mechanism of	- VAAI
	separation, retardation factor, basic	
	principles.	
	Stationary phase, preparation and	
	transfer of the sample, mobile	
	phases and their nature.	
	-	
	Methods of development of the	
	chromatogram, detection methods,	
	physical and chemical.	
	Applications, for determination of	
	purity, following the course of a	
	chemical reaction.	

	UNIT III		15
	UV- Visible Absorption spectroscopy:	Recapitulation of basic concept of spectroscopy. Terms involved in absorption spectroscopy, monochromatic and polychromatic radiation, radiant power, absorbance, transmittance, absorptivity, molar extinction coefficient, wavelength of maximum absorption,	
		Statement of Beer's law & Lamberts' law combined mathematical expression for Beer- Lambert's Law, deviations from Beer-Lambert's law, types of deviations. Components of an optical instrument and their functions, photometers and spectrophotometers.	
		Photometers: Sources, monochromators, sample containers and detectors, block diagram for a single and double beam photometer, (Numerical problems expected.) Photometric titrations	
R	UIA	Basic principles, experimental set up and operational procedures, Requirements for a photometric titration, types of photometric titration curves, and determination of equivalence point. Advantages and limitations	EGE
Expl	ore •	<b>Conductometric titrations</b> Conductometry and conductometric titrations, basic principles, operational procedure, determination of the equivalence point., Conductometric titration curves for the titration of	• Excel
		<ol> <li>Acid –base titrations of all types</li> <li>Mixture of a weak acid and a strong acid vs. strong base and vice versa.</li> <li>Mixture of acid ans salt ag. base</li> <li>Precipitation titrations</li> <li>Complexometric titrations</li> </ol>	

Advantages and limitations.	

# **Paper III – Medical Microbiology**

## Course objectives: -

The objective of this course is to gain insight into Disease Factors and Processes and Diseases Caused by Microorganisms.

**Learning outcomes:** -By the end of the course the student will be able to:

- $\Box$  List the factors playing a role in causing a disease.
- Discuss the various aspects of Systemic Infections including Causative Agents, Symptoms and Prophylaxis.

Gain the technical capability of handling, isolating and identifying various Bacteria.

Course	UNIT	TOPICS	Credits	Lectures
Code				
RUSBTK403	UNIT I Infectious Diseases	Host Parasite Relationship: Normal Flora; Factors Affecting the Course of Infection and Disease; Mechanisms of Infection and Virulence Factors. Infection: Patterns of Infection; Types of Infections; Signs and Symptoms; Epidemiology and Epidemiological Markers.	2	15
<b>R</b> <b>Exp</b>	UNIT II Medical Microbiology Causative Organisms- I	Origin of Pathogens; Vectors; Acquisition of Infection; Koch's Postulates. Skin : S. aureus, S. pyogenes. Respiratory Tract Infections : M. tuberculosis, S. pneumonia (Characteristics Transmission, Course of Infection, Lab Diagnosis, Management of TB, Prevention and Control, Immuno and	GE SEx	<sup>15</sup>
	UNIT III Medical Microbiology - Causative Organisms- II	Chemoprophylaxis, DOTS and MDR Urinary tract infections: <i>E.coli</i> : Characteristics, Virulence, Clinical disease, and <i>E.coli</i> Infections. GI Tract Infections : <i>Salmonella and Shigella spps.</i> (Characteristics, Virulence- Pathogenesis and Immunity, Clinical Disease, Carriers Lab Diagnosis, Phage Typing Prophylaxis and Treatment).		15

Sexually Transmitted Diseases : Syphilis and Gonorrhoea. Nosocomial Infections : Ps. Aeruginosa	

# **Paper IV – Environmental Science**

### Course objectives: -

The objective of this course is to gain awareness about different Types of renewable sources of energy, Xenobiotic compounds and its treatment by bioremediation mechanisms.

Learning outcomes: -By the end of the course the student will be able to:

- □ Gain an understanding of the types of renewable sources of energy and its production.
- □ Study the different xenobiotic compounds and its degradation
- Discuss the various bioremediation strategies.

Course Code	UNIT	TOPICS	Credits	Lectures
RUSBTK404	UNIT I Renewable sources of energy / Waste resources	Sources of available biomass, energy scanarios, Biogas technology- Biogas plant & types, Biodigester. Biogas composition, production and factors affecting production, uses. Biofuels – ethanol production. Microbial hydrogen production, Biodiesel, Petrocrops. Case studies on biogas and biofuel, Advanced biofuels.	2 EG	15
Exp	UNIT II Xenobiotics & waste water treatment	Definition and types of recalcitrant xenobiotic compounds, Hazards from xenobiotics,Biodegradation of xenobiotics, Aerobic waste water treatment Measurement of level of pollution, Process – Preliminary, primary, secondary, tertiary & sludge treatment. Anaerobic treatment of waste water – Microorganisms in sludge treatment.	e o E	<sup>15</sup>
	UNIT III Bioremediation	Concept of Bioremediation. Microorganisms in Bioremediation, Myco remediation and Phytoremediation.		15

Bioremediation T Measuring Bioren Field. Bioaugmer Biostimulation. M Efficacy of Biore	mediation in the ntation and Aonitoring the	

# Paper V – Bioinformatics and Biostatistics

<b>Course objectives:</b> - The objective of this course is learning and understanding basic concepts of Bioinformatics and Biostatistics.				
<b>•</b>		d of the course the student will be al	ble to	
0		ic concepts of Bioinformatics and Bi		
	nd the tools used in	-		
		tools for Analysis of Biological Dat	a	
Course	UNIT	TOPICS	Credits	Lectures
Code			Cicuits	Lectures
	UNIT I	MS Word, PowerPoint, Excel, Coral	2	15
	Introduction to	Draw		
	Computers and	<b>Biological Databases :</b>		
RUSBTK405	Biological	Classification of Databases		
	Databases	based on Resource, Type		
		Of Molecule, Basis Of		
		Information - Raw and		
		Processed Databases;		
		Primary (NCBI, PIR),		
		Secondary (PRINTS,		
	S 15 15 17	OWLS) and Tertiary or	100	1
		Composite (REACTOME, Introduction to KEGG)		
		Databases; Structure and		
		Sequence Databases.		
		Specialized Databases, Protein		_
C W/ HA	ANA A	databank Protein Pattern	- E-	7.00
CXD	lore 🔵	Databases; Protein Structure and		<b>46</b> 21
- MAR		domain classification		<b>VAA</b> I
		(CATH/SCOP).		
		Genome Information Resources:		
		DNA Sequence Databases		
		Specialized Genomic		
		Resources.		
	UNIT II	BLAST:		15
	BLAST &	BLAST and its Types;		
	Structural	Retrieving Sequence using		
	bioinformatics	BLAST, BLAST based		
		searching, FASTA & Dot		
		Plot Method, Introduction		

	to molecular representations on computers, Visualization of biomolecular structures (Protein, DNA, RNA, drugs), Analysis of biological structures. Introduction to Algorithms	
UNIT III Biostatistics	Theory and Problems based on – Coefficient of Correlation and Regression Analysis; Steps in Testing Statistical Hypothesis; Parametric Tests:- Z Test – Single Mean and Two Means, t- Test – Single Mean, Paired and Unpaired; Chi- Square Test.	15

# Paper VI- Molecular Diagnostics

#### Course objectives: -

The objective of this course is learning and understanding Molecular Techniques and utilizing these techniques in Diagnosis.

Learning outcomes: -By the end of the course the student will be able to:

Gain an understanding of the basic Principles used in Molecular Diagnosis.

Gain critical thinking and analytical skills to understand new Diagnostic Methods.

	•	•	•	
		1 * 11 * 1 * 71	1 111 011	1 1 .
	Apply the knowledge and s	skills gained in th	ne course should be useful in	a developing new
		0		
	Diagnostic Kits			
	Diagnostic Kits			
_				

Course	UNIT	TOPICS	Credits	Lectures
Code				
Fyn	UNIT I Basics of Molecular	Introduction to Molecular Diagnostics : Overview of Molecular Diagnostics;	2	15 <b>YCO</b>
RUSBTK406	Diagnostics	History of Molecular Diagnostics; Molecular Diagnostics in Post Genomic Era; Areas used in Molecular Diagnostics; Future Prospects - Commercializing Molecular Diagnostics, Personalized Medicine, Theranostics.		
		Characterization and analysis of Nucleic – Acids and Proteins : Extraction, Isolation and Detection of DNA, RNA and Proteins; Restriction Endonucleases and Restriction Enzyme Mapping.		

UNIT II	Target amplification :		15
Nucleic Acid	PCR - General Principle;		
Amplification	Components of a Typical PCR		
and	Reaction; Experimental Design;		
hybridization	Primer Designing; Control of PCR		
Methods	Contamination and Mispriming;		
	PCR Product Clean-up and		
	Detection. Types of PCR		
	Reverse Transcriptase, Real time,		
	Multiplex & Nested PCR.		
	Hybridization Techniques :		
	Southern, Northern, Western and		
	FISH; Markers, Probes and its		
	Clinical Applications.		
UNIT III	DNA Polymorphism and		15
Molecular	Identification:		_
Biology based	RFLP and Parentage Testing;		
Diagnostics	RFLP and Sickle-Cell Anaemia.		
	Molecular Diagnostics for		
	Infectious Diseases		
	Molecular Testing for Neisseria,		
	Molecular		
	Diagnosis for HIV-1;		
	Genetic Counselling and		
	Molecular Diagnosis		
	Genetic Testing-Need and Uses;		
	genetic Counselling. Case Studies-		
	Diagnostic Testing for Cystic		
	Fibrosis; Fragile X Diagnostic and		
	Carrier Testing.		
	Ethical, Social and Legal Issues to		
	Molecular - Genetic Testing		mark the second s
			7
		7-1	

# Explore • Experience • Excel

## Paper VII- ENTERPRENEURSHIP DEVELOPMENT

### **Objective:**

To develop and systematically apply an Entrepreneurial way of thinking that will allow identification and creation of Business Opportunities.

**Learning Outcome:** By the end of the course the student will be able to:

- Develop an understanding of the systematic process and to select and screen a Business Idea.
- □ Design strategies for setting up successful business ideas.
- □ Creation of unique ideas for business development.

Course	UNIT	TOPICS	Credits	Lectures
Code				
	UNIT I Entrepreneurship Development UNIT II Setting up the Business and Management Aspect UNIT III Innovation and entrepreneurship development	Concept of Entrepreneur; Entrepreneurship; Need and Importance Factors responsible for shaping an entrepreneur. Entrepreneurship development process Difference between entrepreneur and manager. Launching of an enterprise. Enterprise selection, analysis of suitable market, feasibility study, SWOT analysis of business Resource mobilization - financial, technological, raw material. Evaluation of project designing business plan. Principles of management, quality circles, MBO, MBW, TQM Innovation and Marketing Management Marketing management and business development Marketing plan, sales promotion, market segmentation - STP analysis	2 2 5 G e o E	15 15 15
		e		

# **PRACTICALS for SEMESTER IV**

Course Code	Title	Credits
Practicals based on paper 1 and 2	<ul> <li>Determination of blood glucose levels for detection of Diabetes mellitus.</li> <li>Organ Function Tests: Liver (SGPT, SGOT)</li> <li>Kidney function tests (Urea from Serum).</li> </ul>	2
RUSBTKP401	<ul> <li>Qualitative Detection of Ketone Body in Urine.</li> <li>Isolation of Mitochondria and Demonstration of ETC using a Marker Enzyme.</li> <li>Determination of acid number and iodine number of oil/ fats.</li> <li>To determine dissociation constant of weak acid by incomplete titration method using pH meter.</li> <li>Qualitative Analysis of bi-functional organic compounds (minimum four) on the basis of</li> <li>a. Preliminary examination</li> <li>b. Solubility profile</li> <li>c. Detection of elements C, H, (O), N, S and X.</li> <li>d. Detection of functional groups</li> <li>e. Determination of physical constants (M.P/B.P)</li> </ul>	
RU	f. Confirmatory tests to be performed. Solid or liquid Compounds containing not more than two functional groups from among the following classes may be given for analysis: Carboxylic acids, phenol, carbohydrates, aldehydes, ketones, ester, amides, nitro, anilides, amines, and alkyl and aryl halides.	
Explore	<ul> <li>Chromatography:</li> <li>a) Separation of cations: Fe(III), Ni(II) and Cu(II) in a sample by paper chromatography</li> <li>b) Separation of a mixture of o-and p-nitrophenols by thin layer chromatography (TLC).</li> <li>To determine partition coefficient of iodine between water and CCl<sub>4</sub></li> <li>To verify Beer Lamberts law.</li> <li>Determination of amount of Fe (III) in the given solution by photometric titration using salicylic acid.</li> <li>Determination of amount of Fe (II) present in the given solution titrimetrically using diphenylamine indicator.</li> </ul>	cel
Course code	Title	Credits
Practicals based on Paper 3and 4	<ul> <li>Identification of <i>S.aureus</i>-Isolation, Catalase, Coagulase Test.</li> <li>Identification of <i>E.coli</i>-Isolation, Sugar Fermentations, IMViC.</li> </ul>	2

RUSBTKP403	<ul> <li>Identification of <i>Pseudomonas</i> - Isolation, Urease test, Oxidase Test, TSI Slant.</li> <li>RPR Test (Kit Based).</li> <li>Permanent Slide- <i>Mycobacterium</i>.</li> <li>Determination of total solids from an effluent sample.</li> <li>Study of physico-chemical (pH, color, turbidity, BOD, COD) parameters of any one industrial effluent sample.</li> <li>Most Probable Number (MPN) – Presumptive, Confirmed and Completed Tests.</li> <li>Bioremediation of Metal.</li> <li>Mass and energy calculation for biogas and biofuels</li> <li>Visit to STP / CETP</li> </ul>	
Course code	Title	Credits
Practicals based on Paper 5 and 6	<ul> <li>Familiarization with NCBI, EMBL, DDBJ, PIR, KEGG Databases.</li> <li>Use of NCBI BLAST Tool.</li> </ul>	2
RUSBTKP405	<ul> <li>Classification of Proteins using CATH/SCOP.</li> <li>Visualization of proteins using Rasmol.</li> <li>Visualization of biomolecular structures – PyMOL, Chimera.</li> <li>Analysis of three dimensional structures – Similarity, interactions.</li> </ul>	
	<ul> <li>Handling and Calibration of Micropipette.</li> <li>Isolation and Detection of RNA from Bacteria and Yeast.</li> <li>RFLP- Kit Based.</li> </ul>	
RU	<ul> <li>Primer Designing through Open Online Source NCBI- BLAST.</li> <li>DNA Amplification – PCR.</li> </ul>	
Fyplore	<ul> <li>Excel based Biostatistics Practicals.</li> <li>Poster of any recently published paper.</li> </ul>	ral
PVLIA	- ryhei leine - ry	

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# **MODALITY OF ASSESSMENT**

#### **Theory Examination Pattern:**

A)

#### Internal Assessment - 40% :40 marks.

Sr No	Evaluation type	Marks
1	One Assignment	20
	(Animations/Presentations/Posters/ Video Making/ Skits/ Written assignments)	
2	One class Test (multiple choice questions or objective )	20
	Total Marks	40

#### B) External examination - 60 %: 60 marks

#### Semester End Theory Assessment - 60 marks

- i. Duration These examinations shall be of 02 hours duration.
- ii. Paper Pattern:

1. There shall be **03** questions each of **20** marks. On each unit there will be one question. All questions shall be compulsory with internal choice within the questions.

0

2. 60% options will be provided.

Questions	Options	Marks	Questions on		
Q.1)A)	Any 5 out of 8	05	Unit I		
Q.1)B)	Any 3 out of 5	15			
Q.2)A)	Any 5 out of 8	05	Unit II		
Q.2)B)	Any 3 out of 5	15			

Q.3)A)	Any 5 out of 8	05	Unit III
Q.3)B)	Any 3 out of 5	15	

#### **Practical Examination Pattern:**

#### (A)Internal Examination:

Heading	Practical I
Test (2 performing	30
practicals)	
RUSBTKP301/401	
RUSBTKP303/403	
RUSBTKP305/405	
Journal	10
Total	40

Note- Similar pattern for internal practical will be followed for all three Practical papers.

#### (B) External (Semester end practical examination):



Note – Similar pattern for external Practical will be followed for all three practical papers

#### PRACTICAL BOOK/JOURNAL

The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination.

In case of loss of Journal and/ or Report, a Lost Certificate should be obtained from Head/ Co-ordinator / In charge of the department; failing which the student will not be allowed to appear for the practical examination.

#### **Overall Examination and Marks Distribution Pattern**

Course	RUSBTKP301/401			RUSBTKP302/402		Grand Total	
	Internal	External	Total	Internal	External	Total	
Theory	40	60	100	40	60	100	200
Practicals	20	30	50	20	30	50	100

#### Semester: III/ IV

Course	RUSBT	(P303/403		RUSBTKP304/404		Grand	
						CI.	Total
	Internal	External	Total	Internal	External	Total	-A.,
Theory	40	60	100	40	60	100	200
Practicals	20	30	50	20	30	50	100

Course	RUSBTKP305/405			RUSBTKP306/406			Grand Total
	Internal	External	Total	Internal	External	Total	
Theory	40	60	100	40	60	100	200
Practicals	20	30	50	20	30	50	100

