S. P. Mandali's Ramnarain Ruia Autonomous College

(Affiliated to University of Mumbai)



Syllabus for

Program: F.Y.B.Sc.

Program Code: (RUSZOO)

(As per the guidelines of National Education Policy 2020-Academic year 2023-24)

(Choice based Credit System)



GRADUATES ATTRIBUTES

S. P. Mandali's Ramnarain Ruia Autonomous College has adopted the Outcome Based Education model to make its science graduates globally competent and capable of advancing in their careers. The Bachelors Program in Science also encourages students to reflect on the broader purpose of their education.

GA	GA Description
	A student completing Bachelor's Degree in Science program will be able to:
GA 1	Recall and explain acquired scientific knowledge in a comprehensive
	manner and apply the skills acquired in their chosen discipline. Interpret
	scientific ideas and relate its interconnectedness to various fields in
	science.
GA 2	Evaluate scientific ideas critically, analyse problems, explore options for
	practical demonstrations, illustrate work plans and execute them, organise
	data and draw inferences.
GA 3	Explore and evaluate digital information and use it for knowledge
	upgradation. Apply relevant information so gathered for analysis and
	communication using appropriate digital tools.
GA 4	Ask relevant questions, understand scientific relevance, hypothesize a
	scientific problem, construct and execute a project plan and analyse
	results.
GA 5	Take complex challenges, work responsibly and independently, as well as
	in cohesion with a team for completion of a task. Communicate effectively,
	convincingly and in an articulate manner.
GA 6	Apply scientific information with sensitivity to the values of different
	cultural groups. Disseminate scientific knowledge effectively for upliftment
	of the society.
GA 7	Follow ethical practices at the workplace and be unbiased and critical in the
	interpretation of scientific data. Understand the environmental issues and
	explore sustainable solutions for it.
GA8	Keep abreast with current scientific developments in the specific discipline
	and adapt to technological advancements for better application of scientific
	knowledge as a lifelong learner.



PROGRAM OUTCOMES

РО	Description
	A student completing Bachelor's Degree in Science program in the
	subject of ZOOLOGY will be able to:
PO 1	Identify the major groups of organisms, discuss the basis of their biodiversity,
	and draw parallels with their phylogenetic relationship, using well-thought
	cardinal features of classification on the basis of morphology and molecular information.
PO 2	Understand and analyse the evolutionary link amongst the animals and also
	understand the basic classification patterns of invertebrates and vertebrates.
	They will be able to compare and contrast the anatomy and physiology of
	different invertebrates and vertebrate phylum.
PO 3	Analyse the genes, genomes, cells, cell organelles, tissues and histological
	studies, understand the linkage of genes, mechanisms of sex determination,
	various structures of DNA and apply the knowledge of genetics to the process
	of evolution.
PO 4	Analyse and understand the broad concepts of ecology, food webs, food
	chains and the interconnectedness of biotic and abiotic factors. Comprehend
	the concepts of Population dynamics, communities and its dependence on
	the ecosystems.
PO 5	Objectively understand and evaluate information about animal behaviour and
	ecology encountered in our daily lives.
PO 6	Students will be able to demonstrate proficiency in the experimental
	techniques and methods of analysis appropriate for their area of specialization
	within Zoology.
PO 7	Get a flavor of research by working on project besides improving their writing
	skills. It will further enable the students to think and interpret individually.



CREDIT STRUCTURE BSc

Semeste r	Subje	ct 1 DS E	Subjec	GE/ OE course (Across disciplines	Vocational and Skill Enhancemen t Course (VSC) & SEC	Ability Enhancement Course/ VEC/IKS	OJT/FP/CEPCC , RP	Total Credit s
1	4		4	4 (2*2)	VSC-2 + SEC -2	AEC- 2 (CSK) + VEC- 2 (Env Sc.) + IKS-2	, S	22
2	4		4	4 (2*2)	VSC-2 + SEC-2	AEC-2 (CSK)+ VEC-2 (Understandin g India)	CC-2	22
Total	8		8	8	8	10	2	44
Exit opt	Exit option: award of UG certificate in Major with 44 credits and an additional 4 credit Core NSQF course/ Internship or Continue with Major and Minor						NSQF	
3	Majo r 8	0	Minor 4	2	VSC-2	AEC-2 MIL	FP -2, CC-2	22
4	Majo r 8		Minor 4	2	SEC-2	AEC-2 MIL	CEP-2, CC-2	22
Total	16		8	4	4	4	8	44

Exit option: award of UG Diploma in Major with 88 credits and an additional 4 credit Core NSQF course/ Internship or Continue with Major and Minor



	5	DSC 12	DS E 4	Minor 2		VSC-2		CEP/FP-2	22
	6	DSC 12	DS E 4	Minor 2				OJT-4	22
	Total	24	8	4		2		6	44
		Exit	option:	award of	_	in Major with Honours/ Res	132 credits or Cor earch	ntinue with	
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Course Code: RUSSECZOO.E111

Course Title: LABORATORY PRACTICES

Type of Course: Skill Enhancement Courses (SEC)

Academic year 2023-24

COURSE OUTCOMES:

COURSE	DESCRIPTION
OUTCOME	A student completing this course will be able to:
CO 1	Recall good laboratory practices and work safely in the department Laboratory.
CO 2	Identify the different Lab safety symbols and describe them.
CO 3	Use of appropriate glassware.
CO 4	Explain the principle and working of various basic laboratory instruments like colorimeter, microscope, pH meter.
CO 5	Analysis the data and prepare correct graphical presentation for it.
CO 6	Calculate the concentration of solutions, Rf values, central tendencies of group and ungroup data.
CO 7	Prepare simple solution of different concentration
CO 8	Trained In technical skills for further studies



DETAILED SYLLABUS

Course	Unit	Course Title - LABORATORY PRACTICES	Credits/
Code			Hours
	Unit I	1.1. Laboratory safety: a) Introduction to good laboratory practices Concept b) Types of hazards, c) symbols and d) Precautions 1.2. Laboratory glassware a) Pipettes b) Conical flask c) Beakers d) Volumetric flask e) Measuring cylindrical f) Test tubes 1.3. Units of measurement: a) Temperature scale and interconversion: Celsius, Fahrenheit, Kelvin b) Calculations and related conversions: Metric system- length (meter to micrometer) Weight (gram to microgram) Volumetric (Cubic measures) Problem solving based conversions c) Concentrations Percent solutions ppt, ppm, ppb Normality Molarity Molarity Molality Problem solving based solution preparation 1.4. Laboratory Instrument a) Microscope b) Colorimeter c) PH meter	
		1.5.Biostatistics Introduction and scope	
		a) Sampling and its types	
		and to typoo	



c) Central Tendencies (mean, median, mode) d) Tabulation and Graphical representations (Histograms, bar diagrams, pie diagrams)	
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Practical

	Course Code: RUSSECZOOP.E111
Sr. No.	Practical Title- LABORATORY PRACTICES
1.	Problem based unit conversions.
2.	Problem based solution preparation.
3.	Handling of glassware used in Lab
4	Interpretation of safety symbols (toxic, corrosive, explosive, flammable, skin irritant, oxidizing, compressed gases, aspiration hazards and Biohazardous infectious material, Radioactivity, Environmental toxicity.
5.	Study of Central tendencies and plotting of Bar diagram, histogram and pie diagram.
6.	 a) Study of parts of microscope and their functions. b) Technique of focusing a permanent slide under 10X and 45X.
7.	a) Dilution of given sample and estimation of OD using colorimeter
•	b) Calculation of concentration from the given OD using formula.
8.	Calculation of pH of three different samples and confirming the result with pH meter
9.	Separation of amino acids from the mixture by paper chromatography.



References:

- 1. Upadhyay and Upadhyay, Biophysical Chemistry, Himalaya Publishing House, 2009
- 2. Irwin H Segal Biochemical Calculation, Wiley & Sons Inc, 1976
- 3. Introduction to Practical Biochemistry, Tata McGraw Hill Publishing Co. Ltd.
- 4. Statistical Methods In Biology N.T.J Bailey
- 5. Experimental biology (measurement and analysis)- R. H. Kay
- 6. Elements Of Mathematical Biology -Alfred J .Log ka
- 7. Introduction to Practical Biochemistry, Tata McGraw Hill Publishing Co. Ltd.
- S.K. Sawhney and Randhir Singh. Introductory Practical Biochemistry, Narosa Publishing House.
- 9. B. K. Mahajan. Methods in Biostatistics, Jaypee Publications.
- 10. V. K. Sharma. Microscopy and Cell Biology, Tata McGraw Hill Publishing Co. Ltd.
- 11. L. Veerakumari. Bioinstrumentation, M.J.P. Publishers.
- 12. Keith Wilson and John Walker. Principles and Techniques of Practical Biochemistry, Cambridge University Press.
- 13. A. H. Patel. A Manual of Medical Laboratory Technology, Navneet Prakashan Ltd.
- 14. Dr. P. K. Bajpai. Biological instruments and methodology, S. Chand company Ltd.
- 15. Frank H. Stephenson. Calculations in Molecular biology and Biotechnology, Academic Press.



Modality of Assessment: Skill Enhancement Courses (SEC) (1 Credit Theory Course for BSc)

A) Internal Assessment- 40%- 30 Marks

Sr No	Evaluation type	Marks
2	Class Test/Assignment/Open Book Test	10
	TOTAL	10

- B) External Examination (Semester End)- 60%- 45 Marks Semester End Theory Examination:
- 1. Duration The duration for these examinations shall be of 30 Minutes.
- 2. Theory question paper pattern:

Paper Pattern:

Question	Options	Marks	Questions Based on
1	A) (OR) B) (i and ii)	8 OR 8(4+4)	
2	В)	7	UNIT 1
	TOTAL	15	

Practical Examination Pattern: Total Marks 50

C. Internal Examination: 40%- 20 Marks

Heading	Practical
Journal	05
Lab Participation	05
Lab work/ Field report/ Presentation	10
Total	20

D. External Examination: 60%- 30 Marks

Duration – The duration for these examinations shall be of 2 hours.

Semester End Practical Examination:

Particulars	Practical
Major Experiment and/or	30
Minor Experiment,	
Identification and Viva voce	
Total	30