Resolution Number: AC/II (20-21).2.RUS20

S.P. Mandali's Ramnarain Ruia Autonomous College

(Affiliated to University of Mumbai)



Syllabus for Semester V & VI

Program: B.Sc. Non-Conventional Energy Sources and Waste Recycling

Program code: RUSACNCE

(Credit Based Semester and Grading System with effect from the academic year 2023-24)



PROGRAM OUTCOMES

S. P. Mandali's Ramnarain Ruia Autonomous College has adopted the Outcome Based Education model to make its science graduates globally competent and capable of advancing in their careers. The Bachelors Program in Science also encourages students to reflect on the broader purpose of their education.

A student completing Bachelor's Degree in Recall and explain acquired scientifi	Science program will be able to: ic knowledge in a comprehensive manner and
Recall and explain acquired scientif	ic knowledge in a comprehensive manner and
PO 1 apply the skills acquired in their cl	nosen discipline. Interpret scientific ideas and
relate its interconnectedness to various	ous fields in science.
Evaluate scientific ideas critically, a	nalyse problems, explore options for practical
PO 2 demonstrations, illustrate work plan	ns and execute them, organise data and draw
inferences.	XV
Explore and evaluate digital inform	nation and use it for knowledge upgradation.
PO 3 Apply relevant information so gat	hered for analysis and communication using
appropriate digital tools.	
PO 4 Ask relevant questions, understand	scientific relevance, hypothesize a scientific
problem, construct and execute a pro-	pject plan and analyse results.
Take complex challenges, work re	esponsibly and independently, as well as in
PO 5 cohesion with a team for comp	etion of a task. Communicate effectively,
convincingly and in an articulate ma	nner.
PO 6 Apply scientific information with se	nsitivity to values of different cultural groups.
	fectively for upliftment of the society.
Follow ethical practices at work place	ee and be unbiased and critical in interpretation
PO 7 of scientific data. Understand the	environmental issues and explore sustainable
solutions for it.	
Keep abreast with current scientific	c developments in the specific discipline and
PO 8 adapt to technological advancement	s for better application of scientific knowledge
as a lifelong learner.	



PROGRAM SPECIFIC OUTCOMES

PSO	Description					
A studer	A student completing Bachelor Degree in science program with applied component as					
NCE-W	R will be able to:					
PSO 1	Recognize the issues faced by our environment, and the repercussions of its					
	mishandling which they, and their future generations would most suffer from in					
	the coming years.					
PSO 2	Devise suitable actions that are needed to reform the society to avert the					
	environmental crisis					
PSO 3	Develop the sense of rights and responsibilities towards the resources that					
	support us.					
PSO 4	Create awareness in self & the society that, solution to each and every problem					
	begins with oneself.					



PROGRAM OUTLINE

SEMESTER V				
Course Code	Unit	Course Title / Unit Title	Credits	
RUSACNCE501	Non-Conventional Sources of Energy & Waste			
		Recycling		
	NON-CONVENTIONAL SOURCES OF			
	I	ENERGY:		
		Perspectives on Energy Sources and Utilization		
		Study of the following non-conventional energy]	
		sources with respect to scope, present scenario,		
	***	applications, limitations and future prospects:		
	II	Solar Energy		
		Wind Energy		
		Geothermal Energy:		
	WASTE RECYCLING:			
	III	Introduction to waste management.		
		Management of gaseous waste		
	IV	Introduction to liquid waste	1	
	1 1	Particulate matter and its management		
RUSACNCEP501		Practical	2	

SEMESTER VI			
Course Code	Unit Course Title / Unit Title C		
RUSACNCE601	Non-C	onventional Sources of Energy & Waste Recycling	2
	I	Ocean Thermal Energy Conversion Tidal Energy	
II III		Fuel Cells and Hydrogen as a potential fuel	
		Biomass	
		Solid Waste Management	
9	IV	Liquid Waste Management	
RUSACNCEP601	Practical 2		



SEMESTER V

Course Code: RUSACNCE501

Course Title: Non-Conventional Sources of Energy & Waste Recycling Academic year 2020-21

Course Outcomes:

After s	After studying this course, the learner will be able to:				
CO 1	Comprehend the need for energy from the perspective of man, and describe the				
	various energy sources.				
CO 2	Define energy efficiency, and correlate the current energy crisis with the energy				
	efficiency gap				
CO 3	Review different non-conventional energy sources such as solar, wind, geothermal,				
	etc. and assess them critically with respect to feasibility and energy value.				
CO 4	Define the concept of waste, and illustrate the different methods of waste disposal.				
CO 5	Evaluate the parameters to judge the quality of water and management of gaseous				
	waste.				

DETAILED SYLLABUS

RUSACNCE501	Non-C	Credits-02	
	Waste		
	Unit	Unit Unit Title	
	I	Perspectives on Energy Sources and	(15L)
		Utilization:	
		1.1. Definitions, units and concept, energy	
		requirements of man and society, unequal	
		distribution of energy resources around	
		the world, Energy consumption patterns,	
40		Contrast in energy demands of developed,	
		developing and under-developed	
		1.2. Sector wise need for energy: Domestic,	
		industrial, agricultural, and transport	
		sectors. Environmental implications of	
		energy use, Concept of sustainable	
		development and its energy execution,	
U		Energy audit and its types.	
		1.3. Various types of energy sources,	
		conventional/ non-conventional,	



		renewable / non-renewable, fossil fuels:	
		coal, oil, natural gas; thermal,	
		hydroelectric and nuclear.	
	1.4	• Energy storage and conservation methods.	
		Role of government and Non-	
		Governmental Organizations, socio-	
		economic aspects, Government policies.	
	1.5	-	
	1.5	Fundamentals of energy efficiency, the	
		energy efficiency gap, participants in the	
		field of energy efficiency, Energy	
		efficiency policies.	
	1.6	• The climate debate: History of climate	
		policy, clashing positions on climate	
		change: skeptics and supporters, climate	
		realists, case studies on climate change	
		activism around the world. Role of	
		individual on the climate crisis, concept of	
		carbon footprint.	
	1.7	Entrepreneurship opportunities in the	
	1./	energy sector.	
<u> </u>	TT C4-		(71)
		dy of the following non-conventional	(7L)
		ergy sources with respect to scope, present	
		nario, applications, limitations and future	
	-	ospects:	
		Solar Energy	
	2.1	.1 Solar spectrum and its relevance as the	(4L)
		energy source, solar radiation,	,
		conversion of solar energy into heat	
		energy in solar collectors, Fresnel lenses,	(41)
		parabolic reflectors.	(4L)
	2.1	.2 Photovoltaic effect, semi-conductors as	
		solar energy converters, different	
		materials used, factors affecting	
		photovoltaic efficiency, effect of	
		temperature, solar cells.	
	2.1	.3 Photo-electrochemical process for the	
		conversion of solar energy; a pplications	
		of solar energy for different purposes.	
$\sim 0^{\circ}$	2.2	Wind Energy	
	2.2		
	ener	• •	
		2.2 Types of wind mills and their working,	
	2.2		
		rotor with blades, gear generator, vertical	
		and horizontal axis of rotation.	
	2.3	Geothermal Energy:	
ļ.			



	221 0:::	
	2.3.1 : Origin of geothermal heat, temperature	
	gradient, geothermal steam and hot	
	spring, geysers.	
	2.3.2: Power production from geysers, mantle	
	as heat source.	4 1 1
III	WASTE RECYCLING:	(9L)
	3.1 Introduction to waste management:	
	3.1.1 Definition of waste, types of waste,	
	changing trends in waste generation,	
	resources, development, population and	
	waste generation, concept of waste	(6L)
	management and approaches to it: end of	(OL)
	the pipe treatment, in plant treatment, goal	
	of zero waste and its feasibility,	
	Introduction to green chemistry.	
	3.1.2 Waste disposal and its three 'R"s	
	Recovery, reuse, recycle and disposal of	
	waste, economic viability of each of the	
	above, waste audit.	
	3.1.3 Waste generation: Types of waste on the	
	basis of usage, sources, physical state,	
	feasibility with respect to three Rs.	
	domestic, industrial, agricultural and	
	commercial wastes; gaseous, solid and	
	liquid waste.	
	3.2 Management of gaseous waste:	
	3.2.1 : Generation of gaseous waste, smoke, fog	
	mist, emission of gases, Sources,	
	contribution of different sources to total	
	waste. Effects: vehicular and chimney	
	exhaust	
	3.2.2 Treatment methods, adsorption,	
* * * * * * * * * * * * * * * * * * * *	absorption, catalytic conversion, their	
	feasibility cost and regulation	
4' \(\)	3.2.3 Minimization methods, economic	
1 0		
77/	viability of reuse.	(O.T.)
IV	4.1 Introduction to liquid waste	(8 L)
	4.1.1 : Liquid Waste: Generation, local and	
	global sources, variation in the amount	
V),	generated.	
	4.1.2 Characterization of liquid waste	(7 L)
	4.1.3 Physical parameters; colour, odour,	
U	turbidity, TSS. TDS, TS	
	4.1.4 Chemical parameters: pH, acidity,	
>	alkalinity, hardness, DO, COD, TOC,	



THOD and BOD, chemical composition, classification. 4.2 Particulate matter and its management:	
 4.2.1: Particulate matter: sources, relation to energy consumption, characterization, effects. 4.2.2 Disposal methods, cyclone separator, wet scrubber, electrostatic precipitator, fabric filter. 4.2.3 Regulations and their implementation in case of waste disposal, Role of the government. 	Olle

Semester-V Practical

DUCA CNICEDEA1		C . 1'4 . 02	
RUSACNCEP501		Credits-02	
		Recycling	
	1.	Determination of acidity, alkalinity and hardness of	
		the given water sample.	
	2.	Construction of the breakthrough curve and	
		determination of breakthrough capacity of the given	
		ion exchanges resin.	
	3.	Determination of T.S., T.S.S. and T.D.S. present in	
		the given water sample.	
	4.	Estimation of chloride in water sample by	
		argentometric method	
	5. Determination of nitrite colorimetrically from the		
. ?	waste water sample.		
	6. Determination of sulphate in the given water sample		
~(0,		by benzidine sulphate method.	
	7.	Determination of Cr (VI) spectrophotometrically in	
		the given water sample.	
	8.	Determination of phosphates in water by stannous	
5		chloride method.	
	(2 6	experiments to be done at the time of examination)	



Modality of Assessment

Theory Examination Pattern:

A) Internal Assessment – 40 % (40 marks).

Sr No	Evaluation type	Marks
1	One class Test (multiple choice questions / objective)	20
2	Seminar Presentations/ Assignments	15
3	Active participation in routine class activities	05
	Total	40 Marks

B) External examination - 60 %

Semester End Theory Assessment - 60%

60 marks

- i. **Duration -** These examinations shall be of **two hours** duration.
- ii. Theory question paper pattern: -There shall be **four** questions each of **15** marks, one on each unit. All questions shall be compulsory with internal choice within the questions.

Questions	Options	Marks	Questions on
Q.1) a)	Any 3 out of 5	12	I Init I
Q.1) b)	Any 1 out of 2	03	Unit I
Q.2) a)	Any 3 out of 5	12	11
Q.2) b)	Any 1 out of 2	03	Unit II
Q.3) a)	Any 3 out of 5	12	I I:4 III
Q.3) b)	Any 1 out of 2	03	Unit III
Q.4) a)	Any 3 out of 5	12	I I:4 IV
Q.4) b)	Any 1 out of 2	03	Unit IV

Practical Examination Pattern:

A)Internal Examination: - 501

Activity	Marks
Internal of Practical Test	10
Laboratory work	15
Biogas Plant Maintenance or site visit with relevant reports	15
Total	40



B) External Examination 501 (Semester end practical examination):-60 Marks

Sr. No.	Particulars	Marks
1	Laboratory Work	25 + 25
2	Viva and Journal	5+5
	Total	60

PRACTICAL BOOK / JOURNAL:

The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination.

In case of loss of Journal and/ or Project Report, a Lost Certificate should be obtained from Head/ Co-ordinator / In charge of the department; failing which the student will not be allowed to appear for the practical examination.

Overall Examination and Marks Distribution Pattern

Course	Į.		
	Internal	External	Total
Theory	40	60	100
Practicals	40	60	100

(Total Marks: 200)



Semester - VI Course Code: RUSACNCE601

Course Title: Non-Conventional Sources of Energy & Waste Recycling Academic year 2020-21

Course Outcomes:

After s	tudying this course, the learner will be able to:
CO 1	Summarize the ocean-based non-conventional energy sources- namely OTEC and
	Tidal, and appraise them in keeping with the current local and global scenario.
CO 2	Explain the construction and working of fuel cells, and recognize their potential as
	the solution to the global energy crisis
CO 3	Explore the prospects of hydrogen as a fuel.
CO 4	Define biomass, describe the use of biomass as an energy source.
CO 5	Outline the use of biomass in India.
CO 6	Characterize the management of solid waste with respect to its sources, and disposal
	methods.
CO 7	Appreciate the significance of recycling, reuse & reclamation of solid waste.
CO 8	Describe the various treatment methods used for three step (physical, chemical and
	biological processes) treatment of liquid waste in general, applications in specific
	industries like fertilizer, food & beverage, petrochemical industry.
CO 9	Examine the government policies and regulations with regarding consumable and
	non-consumable water.



DETAILED SYLLABUS

D	Non	a	
RUSACNCE601		Credits-02	
	Unit	Waste Recycling Unit Title	Lectures
	I	Study of the following non-conventional	(3L)
		energy sources	
		1.1 Tidal Energy:	
		1.1.1 :Tide height and its potential for power	
		generation.	(6L)
		1.1.2: Construction of dam across sea basin,	
		installation of the turbine, specification	
		of the basin.	
		1.2. Ocean Thermal Energy Conversion	(3L)
		1.2.1 Difference in surface temperatures and	
		at a depth in ocean, its use as a source	
		for power generation	(3L)
		1.2.2 Requirements for a practical OTEC	
		plant, different working fluids.	
		1.2.3 Relevance to the present scenario,	
		feasibility and future projection.	
		1.3. Fuel Cells:	
		1.3.1 Electrochemical energy conversion,	
		basic principle of fuel cells, advantages	
		of energy conversion in fuel cells,	
	\triangle	distinction between fuel cell and battery.	
		1.3.2 Hydrogen – oxygen fuel cell, organic	
		oxygen fuel cell	
		1.3.3 Applications of fuel cells.	
		1.4 Use of hydrogen as a potential fuel:	
		availability, generation and use,	
(O.		feasibility and economic viability,	
		advantages, Economically viable	
~0	TT	hydrogen production, photolysis of water.	(151)
	II	Biomass:	(15L)
		2.1 Solar energy and generation of biomass, Bio-energy: Biomass as a resource,	
		Advantages of biomass energy among	
		alternative resources, opportunities and	
		challenges, forms of biomass: solid,	
		liquid, gas.	



		 2.2 Biomass properties, types and energy value, utilization with and without conversion. 2.3 Importance of lignin with respect to biomass energy, lignin content in various biomass, Non-lignocellulosic biomass (algae) and microalgae cultivation, macroalgae (seaweed) cultivation. 2.5 Biomass processing methods: Physical, thermochemical (pyrolysis, torrefaction, gasification, trans-esterification), biochemical (anaerobic digestion, fermentation, enzymatic conversion). 2.6 Biogas and biofuels: Production and case studies on its use. Contribution of biomass-based energy resources on India's energy use, Role of government, policies introduced as incentives for furthering biomass energy resources, role 	
		furthering biomass energy resources, role of NGOs.	
	III	WASTE RECYCLING:	(15L)
		3.1 Solid Waste Management:	
		3.1.1 Solid waste, sources and their	
		characterization.	
		3.1.2 Classification of solid waste: chemical nature, biodegradable and non bio-	
		degradable, their feasibility with respect	
		to the use of three 'R's. 3.1.3 Feasibility of recycling: metal, paper,	
		plastic rubber and glass, costs and	
		economic viability, need of the	
•		treatment at the source level, role of the	
		government and awareness in citizens.	
A* O	•	3.1.4: Methods of disposal of solid waste:	
		3.1.4.1 Dumping of garbage,	
		3.1.4.2 Sanitary landfills,	
10		3.1.4.3 Composting, soil conditioning, vermi-	
ΔI_{i}		composting.	
		3.1.4.4 Incineration;	
		3.1.5 Hazardous and toxic waste, definition,	
()·		classification, difference between the	
		hazardous and toxic waste, methods of	
		minimization/	



	3.1.6 Radioactive waste: sources, effects on	
	plants, animal and man, activity level	
	and its management, minimization and	
	treatment.	
	3.1.7 e-waste: types, effects and management.	
	3.1.8 Hospital and medical waste, disposal	
	and preventive measures.	
	3.1.9 Disaster management and risk analysis,	
	restriction of hazardous substances.	
IV	Liquid Waste Management:	(15L)
	4.1 Waste water, industrial effluent, need for	
	the treatment. sources,	
	4.2 Use of COD, BOD and TOC for deciding	
	the treatment process.	
	4.3 Pre-primary treatment: neutralization,	
	equalization.	
	4.4 Primary treatment: screening,	
	sedimentation, coagulation, filtration	
	4.5 Secondary treatment: principles of the	
	biological treatment of liquid waste,	
	4.6 Various processes used: aerobic and	
	anaerobic process.	
	4.6.1 Aerobic process, oxidation ponds,	
	oxidation ditch, Aerated lagoons,	
	activated sludge process and trickling	
	filter process.	
	4.6.2 Anaerobic processes, anaerobic contact	
	process	
	4.7 Tertiary treatment: reverse osmosis,	
	ultrafiltration, electro-dialysis, ion	
	exchange, ozone treatment.	
	4.8 Government regulations, permissible	
	levels for drinking water and for other	
40	uses.	
	4.10 Characterization of effluent from i)	
	Petrochemical ii) food and beverage iii)	
	Fertilizer industry.	
1	-	



Semester-VI Practical

RUSACNCEP601	No	n-Conventional Sources of Energy & Waste Recycling	Credits-02
	1.	Determination of COD of the waste water sample.	1/10
	2.	Determination of dissolved oxygen present in the	
		given water sample.	0
	3.	Determination of dosage of coagulant by Jar Test.	
	4.	Determination of iron in water sample by	
		phenanthroline method / AAS method	
	5.	Group Activities:	
	6.	Preparation of Biofuel from waste oil	
	7.	Soil analysis.	
	8.	Introduction to survey methodology, designing	
		mock surveys in relation to non-conventional energy	
		sources and waste recycling.	

REFERENCE BOOKS

- 1.Solar energy: principles of thermal collection and storage S.P. Sukhatme, Tata Mcgraw Hill, New Delhi 1990.
- 2. Fuel Cell Will Mitchell, Academic Press 1963.
- 3.Photo electrochemical Cells: Studies in Physical and theoretical Chemistry, Vol. 50 K.V.S.Santham & M.Sharma, Elsevier Publishing Company, Amsterdam, 1988.
- 4. Wastewater Treatment and pollution control: S.A.Arsewala, Tata Mcgraw Hill, New Delhi 1990.
- 5. Pollution Control in Process Industries: S.P. Mahajan, Tata Mcgraw Hill, New Delhi 1990.
- 6. Waste Water Treatment: M.N. Rao and A.K. Datta, Oxford and IBH Publishing.



Modality of Assessment

Theory Examination Pattern:

A) Internal Assessment -40 % (40 marks).

Sr No	Evaluation type	Marks
1	One class Test (multiple choice questions / objective)	20
2	Seminar Presentations/ Assignments	15
3	Active participation in routine class activities	05
	Total	40 Marks

B) External examination - 60 %

Semester End Theory Assessment - 60%

60 marks

i)Duration - These examinations shall be of two hours duration.

ii)Theory question paper pattern: -There shall be **four** questions each of **15** marks, one on each unit. All questions shall be compulsory with internal choice within the questions.

Questions	Options	Marks	Questions on
Q.1) a)	Any 3 out of 5	12	Unit I
Q.1) b)	Any 1 out of 2	03	Omt 1
Q.2) a)	Any 3 out of 5	12	Unit II
Q.2) b)	Any 1 out of 2	03	Omt II
Q.3) a)	Any 3 out of 5	12	Unit III
Q.3) b)	Any 1 out of 2	03	Ollit III
Q.4) a)	Any 3 out of 5	12	Unit IV
Q.4) b)	Any 1 out of 2	03	Unit IV

Practical Examination Pattern:

A) Internal Examination: - 601

Activity	Marks
Internal of Practical Test	10
Laboratory work	15
Biogas Plant Maintenance or	15
site visit with relevant reports	13
Total	40



B) External Examination 601 (Semester end practical examination):-60 Marks

Sr. No.	Particulars	Marks
1	Laboratory Work	25
2	Viva and Journal	5+5
3	Project work	25
	Total	60

PRACTICAL BOOK / JOURNAL:

The students are required to present a duly certified journal for appearing at the practical examination, failing which they will not be allowed to appear for the examination. In case of loss of Journal and/orProject Report, a Lost Certificate should be obtained from Head/Co-ordinator / In charge of the department; failing which the student will not be allowed to appear for the practical examination.

Overall Examination and Marks Distribution Pattern

Course	601		
	Internal	External	Total
Theory	40	60	100
Practicals	40	60	100

(Total Marks: 200)